TARGISOL hands on session.

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Abstract

Diffusion is often the dominant mechanism in the release of radioactive isotopes. DIFFUSE is a finite elements application that allows to make diffusion calculations under different regimes and with variable profiles and with time structures. DIFFUSE is part of the RIBO project.

1 Diffusion

- 1. The activity of radioactive atoms implanted in different matrices may be monitored to evaluate the diffusion coefficients of those isotope-matrix pairs. Let us imagine that 1.37 % of a certain isotope ^AX of half-life $T_{1/2} = 5 s$ remains inside a 100 μm foil 30 s after the starting of the measurement.
 - (a) What would remain if it were a stable nuclei? *A- obviously* $(2^{30/5} = 2^6) \cdot 1.37 \% = 87.6672 \%$
 - (b) What would then be the diffusion coefficient? (consider a flat starting profile)
 - (c) Check the correctness. Find 1.37 % by using the first law of Fick including radioactive decay.
 A- use 1(S) /2(3) /4(Rad...) /9(100) /10(100) /11(1E-8) /13(30) /14(600)
- 2. Improving diffusion...
 - (a) Considering only diffusion (not decay). How long would it take to reach the same fraction (87.667 %) if the foil were only 50 μm thick?
 A- use 1(S) /2(1) /3(50) /11(1E-8) /13(30) /14(600) → 7.5 s
 A- or automatically, T = 30 · (50/100)² = 7.5 s
 - (b) What are the limitations to thinning the foils? *A- Effusion, price, sintering.*

- (c) What would be the remaining fraction (no decay) after 30 s from a cylinder of a diameter of 30 μm with the same diffusion coefficient. And from a Sphere?
- 3. Depending on the matrix thickness and the energy of the irradiating beam, the starting concentration may be flat or it may show a distribution in space which may be estimated with software like TRIM.
 - (a) Repeat the calculation 1.2. to extract the diffusion coefficient for the following starting distribution:

$$exp\left(\frac{5\cdot(x-a)}{a}\right)\cdot\left(1-exp\left(\frac{(x-a)}{a}\right)\right) \tag{1}$$

a is the thickness of the foil. 6:(exp(5 \star (x-a)/a) \star (1-exp((x-a)/a)) 1(S) /2(5) /4(Fancy) /6(...) /9(100) /10(100) /12(0.87667) /13(30) /14(300) $\rightarrow D = 2.08 \cdot 10^{-8} \text{ cm}^2/\text{s}$

(b) Plot the evolution C(x,t) for $D = 1 \cdot 10^{-8} cm^2/s$ with the peak starting profile and no decay up to 320 s in 40 time bins. What is the tendency of the concentration peak?

A- 1(S)/2(4)/4(Fancy)/6(...)/9(100)/10(100)/11(1E-8)/13(30)/14(300) The distribution becomes symmetric with the peak in the center which smooths down progressively.

- (c) Compare it with the C(x,t) function for the flat starting concentration.
 A- 1(S)/2(4)/4()/9(100)/10(100)/11(1E-8)/13(30)/14(300) → D = 2.08 · 10⁻⁸ cm²/s The distribution becomes symmetric with the peak in the center which smooths down progressively.
- (d) If the atoms stick in the surface for a relatively long time, the concentration in the surface of the foils increases. Does that favor diffusion from the foil?
- 4. Diffusion coefficient ...
 - (a) What would be the remaining fraction after 30 s if the foil were heated in such a way that the diffusion coefficient would grow linearly from 0.5 · D₀ to 1.5 · D₀? (flat starting concentration, no decay).
 A- 1(S) /2(3) /4(Fancy Diff...) /8(Dc*(0.5+(x/a)) /9(100) /10(100) /11(1E-8) /13(30) /14(300) → 91.30 %.
 - (b) What would happen if the diffusion coefficient doubled after a rise of temperature?.

A- use $I(S)/2(1)/3(50)/11(2E-8)/13(30)/14(600) \rightarrow 15 s$ A- or automatically, $T = 30 \cdot (10^{-8}/2 \cdot 10^{-8}) = 15 s$

- (c) What are the limitations to rising the temperature in the foils? *A- Sintering, melting.*
- 5. Pulsed beam ...
 - (a) Obtain the diffusion release from a target $(100 \ \mu m \text{ foils}, D = 10^{-8} \ cm^2/s)$ under a pulsed beam of period $T_p = 1 \ s$, for the cases $T_{1/2} = 10, 5, 1s$. A- use $I(S) \ /2(3) \ /4(Rad... Beam...) \ /5(10) \ /7(1) \ /9(100) \ /10(100) \ /10(100) \ /11(1E-8) \ /13(30) \ /14(900)$
 - (b) What can you conclude?

2 Effusion

2.0.1 Effusion